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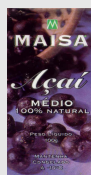
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**Figure 1**  
Açaí palm with fruits

## *Euterpe oleracea* and its products

*Euterpe oleracea* Mart., Arecaceae, is a palm widely spread in the Amazonian estuaries and floodplains of South America. To the local inhabitants the palm is known as "Açaí" and it is currently the world's most important source of palm hearts. The grape-sized and dark-purple fruits (see **Fig. 1**) are used for the creation of several dishes like ice-cream, pies and jelly. The major product is a beverage made from the pulp by diluting it with different amounts of



**Figure 2**  
Açaí liquid

water (see **Fig. 2**); the most important trade qualities are Açaí grosso, medio and fino. Açaí products have a high nutritional value, because lipids can account for up to 50% of the dry matter. Their taste is described as unusual but savoury and they enjoy great popularity throughout northern South America [1, 2]. A launch of Açaí products to the European market is currently under preparation.

## Analysis of antioxidant capacities

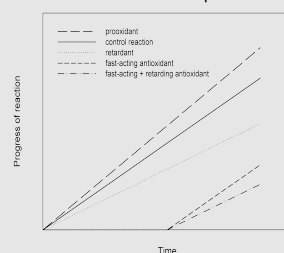
The analysis of antioxidative capacities was carried out with the TOSC (Total Oxidant Scavenging Capacity) assay [3,4] in a modified and automated version [5]. This assay is based upon the ethylene yielding reaction of KMBA ( $\alpha$ -keto- $\gamma$ -methiolbutyric acid) with three reactive oxygen species (peroxyl radicals, hydroxyl radicals and peroxynitrite, respectively). The time course of ethylene production was monitored during one hour by repeated headspace GC with a Combi-PAL autosampler (see **Fig. 3**). The kinetic curve that best fits the experimental data and the area beneath it were calculated. TOSC values were quantified by comparing the areas for



**Figure 3**  
GC with Combi-PAL autosampler (CTC)

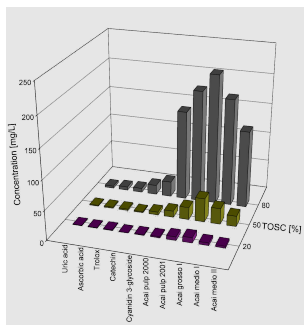
(uninhibited) control and sample reaction: a sample without antioxidative capacity has a TOSC value of 0%, a complete suppression of ethylene formation corresponds to a TOSC value of 100% and prooxidants obtain negative TOSC values. Samples were analysed in at least five different concentrations to cover the TOSC range as complete as possible. Curves were fitted through the experimental TOSC values and the corresponding concentrations. Based on these equations, concentrations were calculated, that match TOSC values of 20, 50 and 80%, respectively. In addition, fast-acting anti-oxidants and retardants [6] were distinguished by kinetic (see **Fig. 4**).

**Figure 4**  
Fast-acting antioxidants, retardants and prooxidants



## Materials

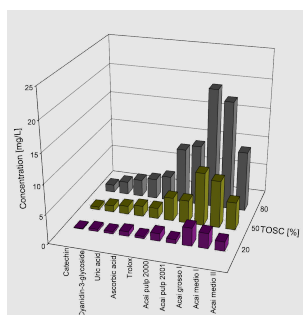
Açaí pulps from 2000 and 2001 were freeze-dried in Brazil directly after preparation. The pulps were suspended with UHQ water and sonicated. Açaí grosso and medio were obtained as commercial products. All samples were centrifuged and filtrated afterwards. The concentration of the resulting solutions was determined by freeze-drying. Diluting of solutions was also made with UHQ water.



**Figure 6** Antioxidative capacities towards peroxynitrite

## Results

All Açaí products were very active against peroxyl radicals (see **Fig. 5**). The inhibition of peroxynitrite and hydroxyl radicals was also significant but lower (see **Fig. 6** and **7**). All undiluted commercial products (Açaí grosso, medio I and II) suppressed the reaction of peroxyl radicals completely. For hydroxyl radicals, this could only be observed for Açaí grosso and medio II. None of the three inhibited peroxynitrite entirely. These findings can be explained by the fact, that Açaí is only for peroxyl radicals a fast-acting antioxidant, while it works as a retardant towards peroxynitrite and hydroxyl radicals (see **Table 1**).



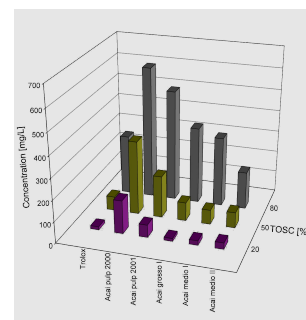
**Figure 5** Antioxidative capacities towards peroxyl radicals

## Acknowledgement

This research was supported by Unilever Bestfoods

	Peroxyl radicals		Peroxynitrite		Hydroxyl radicals	
	fast-acting	retarding	fast-acting	retarding	fast-acting	retarding
Trolox						
Uric acid	+	—				+
Ascorbic acid						+ / proox.
Catechin	—	+		+	—	
Cyanidin-3-glyc.	+	+				+
Açaí pulps	+	+				

**Table 1** Classification of compounds by reaction mode



**Figure 7** Antioxidative capacities towards hydroxyl radicals

## References

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## Thanks

We would like to thank Dr. José G. Maia from the Emilio Goeldi Museum in Belém, Brasil, the Klaus Böcker GmbH from Buxtehude and Mrs Roberta Rodrigues for the friendly supply of Açaí products.